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inclusion bodies, disaggregation can be achieved only in the presence of a strong denaturant such as Gdn-HCl [4].

Allowing for this, we undertook in this work the *in vitro* characterization of the aggregation-disaggregation transition of the α -amylase folding intermediate under conditions partially mimicking the *in vivo* environment.

The fact that α -amylase is an exocellular protein strongly suggests that *in vivo* the unfolded nascent polypeptide chains are transiently in contact with amphipathic compounds such as membrane phospholipids. We therefore used low concentrations of TFE to mimic such a local cellular environment [5].

2. MATERIALS AND METHODS

Purification of exocellular α -amylase

α -amylase was prepared from the culture supernatant of *B. subtilis* GM96101 (*degU32* (Hy), *sacA321*, Δ *sacRsacB*, *sacRamyE*), a derivative of *B. subtilis* 168 [6]. A stock solution of pure protein (60 mg/ml) was prepared in 0.1 M potassium phosphate, at pH 7.

Assay of α -amylase

α -amylase activity was assayed at 37°C, using p-nitrophenylmaltotrioxide as a substrate (bio-Merieux) at pH 6.3 in 0.1 M potassium phosphate. One enzyme unit corresponds to 25 μ g pure α -amylase.

Fluorescence measurements

Changes in intrinsic fluorescence and the fluorescence spectra of α -amylase were recorded with a F2000 Hitachi thermoregulated spectrofluorimeter.

FTIR spectroscopy

We used a Bruker vector 22 equipped with an MCT detector. Spectra were obtained from 100 summed interferograms at a resolution of 4 cm^{-1} for each measurement. Fourier self deconvolution was performed with FD software (Spectrum Square Associates) implemented with GRAMS (Galactic Ind Co) [7]. Fourier derivation was performed using RAMOP software [8]. We estimated the amount of the various secondary structures from the FTIR data by fitting Gaussian peaks at each frequency (see text) to the deconvoluted spectrum and calculating their area relative to the amide I band using GRAMS software.

3. RESULTS

3.1 Competition between folding and aggregation when unfolded α -amylase is at a high concentration and in the presence or absence of TFE

The unfolding-refolding transition of α -amylase at 37°C and pH 7 proceeds *via* two-steps [1] with a partially folded intermediate. This intermediate can be stabilized in a calcium-free refolding medium. This metal ion triggers the conversion to the protein to its native state, and the transition can be monitored by changes in protein intrinsic fluorescence. Previous studies on the unfolding-folding transition used low concentrations of protein ($\approx 0.1 \mu\text{M}$). We reinvestigated the transition with high concentrations of protein ($\approx 1 \mu\text{M}$) and in the presence of low concentrations of TFE. Gradually adding TFE to the refolding medium led to a decrease in the refolding yield (Fig. 1). The α -amylase folding intermediate refolded rapidly, $t_{1/2} = 78 \pm 2$ s, with a yield of 61% in the absence of TFE (Fig. 1A). Refolding was totally inhibited in 10% TFE due to the formation of aggregates, as demonstrated by the distribution of α -amylase between the soluble phase and the protein pellet after centrifugation of the refolding medium (Fig. 1B). There was partial aggregation of the refolding intermediate, even in the absence of TFE. The soluble refolded α -amylase in the supernatant had the same specific enzyme activity as the native protein, while resuspended aggregates were devoid of any enzyme activity. Native α -amylase remained in a soluble and active form when incubated under the same conditions of pH and temperature within the same range of TFE concentrations (not shown). These results therefore indicate that the refolding intermediate tends to aggregate rather than fold when it is at a high concentration and in the presence of an amphipathic compound. Since it was shown from various model proteins that no substantial structural changes occur at these low TFE concentrations [9, 10], it can be postulated that the net effect of this solvent must be to strengthen the interactions between the individual monomeric units involved in aggregates.

3.2 Kinetics of aggregation of α -amylase refolding intermediate

The formation of aggregates from the stabilized folding intermediate I, incubated at 37°C, was detected by an increase in light scattering at 340 nm (Fig. 2). We assumed a constant mean aggregation number, n , for aggregates and postulated that the rate of aggregation is controlled by an initial step that has a reaction order α . The differential rate of the aggregation reaction is thus:

$$\frac{d(nI)}{dt} = k(a - nI)^\alpha$$

where k is the apparent rate constant of aggregation and a the initial concentration of the

folding intermediate. The integrated form of this equation is

$$kt = \frac{1}{\alpha - 1} \left[\frac{1}{(\alpha - nI)^{\alpha-1}} - \frac{1}{a^{\alpha-1}} \right]$$

and

$$t_{1/2} = \frac{2^{\alpha-1} - 1}{(\alpha - 1)ka^{\alpha-1}}$$

The value of $(a - nI)$, which is the concentration of the folding intermediate that remains unaggregated at any time during aggregation, is proportional to the difference in absorbance $\Delta A_t = A_\infty - A_t$.

The kinetic parameters α and $t_{1/2}$ were evaluated (Table 1) by fitting the curves in Fig. 2. The value of the reaction order (α) of the aggregation reaction is close to 2, implying that the rate of aggregation is controlled by an initial dimerization step.

The kinetics of the aggregation in 15 % TFE was monitored with respect to the initial concentration of the folding intermediate. As expected for the reaction with a reaction order > 1 , the half-time of the process decreased with an increase in the initial concentration of the unfolded protein (Fig. 3A). The kinetic parameters of aggregation were evaluated from a plot of $\ln t_{1/2}$ versus $(\ln a)$ (Fig. 3B). The parameters obtained were: $\alpha = 1.80 \pm 0.05$ and

$$k = 0.46 \pm 0.02 \mu\text{M}^{-1} \text{min}^{-1}.$$

3.3 Physico-chemical properties of the aggregates of α -amylase refolding intermediate

FTIR spectroscopy

Estimates of the secondary structure content of the native α -amylase and of the stabilized refolding intermediate indicated that the latter was partially folded [1]. We therefore investigated whether the secondary structure content was preserved or modified in the aggregated state.

The FTIR spectroscopy results indicated that the native α -amylase and the aggregated form of its folding intermediate had somewhat different secondary structures (Fig. 4, Table 2). Most of the information on the protein's secondary structure is contained in the amide I band [7]. Resolution-enhancement procedures such as Fourier self-deconvolution and Fourier-derivation were used to resolve the components (Fig. 4B) underlying the broad amide I band contour because of marked overlap in the spectra (Fig. 4A). The components revealed by the two methods were similar. There were three main components at 1687, 1652 and 1633 cm^{-1} for the native protein and three main

components at 1685, 1641 and 1618 cm^{-1} for the aggregates of the refolding intermediate.

The native protein mainly had components assigned to β -sheets (with a low frequency component at 1633 cm^{-1} and a high frequency component at 1687 cm^{-1}) and to α -helices (1652 cm^{-1}). Fitting the component bands to the amide I contour gave an estimate of the relative amount of secondary structure (Table 2).

The aggregates of the refolding intermediate had a strong component at 1618 cm^{-1} . This low frequency component, associated with the high frequency found at 1685 cm^{-1} , is assigned to aggregated structures such as intermolecular β -strands [11]. The band at 1641 cm^{-1} is broad, suggesting that it is made up of smaller overlapping, unresolved bands induced not only by β -sheets (1648 cm^{-1}), but also by disordered structures and possibly α -helices (1656 cm^{-1}) (Table 2). Comparison with the secondary structure content of the soluble refolding intermediate evaluated previously from far U.V. CD spectroscopy [1] indicated that aggregation led mainly to an increase in the amount of β -sheet.

Resistance to proteolysis

We have demonstrated that the refolding intermediates of α -amylase [1] are sensitive to proteases, in contrast to the native form, but there is no evidence of this when the refolding intermediate is an aggregate. We examined the possibility of kinetic competition between aggregation and proteolytic degradation, postulating that such competition could be important *in vivo*, since there are many proteases associated with the *B. subtilis* envelope.

The first part of this work shows that the rate of aggregation of the refolding intermediate is greatly increased by TFE. We went on to examine this competition over a range of TFE concentrations. The results indicated that the refolding intermediate does not undergo proteolytic degradation when it is very rapidly aggregated (Fig. 5).

3.4 Effectors of disaggregation

The strong denaturing agent Gdn-HCl has been used to solubilize inclusion bodies resulting, *in vivo*, from the irreversible aggregation of unfolded overexpressed proteins. We established the curve describing the solubilization of α -amylase from aggregates by Gdn-HCl, at pH 7 and 37°C, by measuring the protein in the soluble and insoluble phases after incubating the aggregates in various concentrations of denaturant. Concentrations of Gdn-HCl above one molar caused abrupt disaggregation (Fig. 6). This could reflect the cooperative nature of the interactions stabilizing the

aggregated state. $D_{1/2}$, the mid-point of the solubilization curves was approximately 1.25 M. This indicates that the intermolecular interactions are weaker in these aggregates than in inclusion bodies, which are solubilized by higher Gdn-HCl concentrations. Therefore, in order to test the possible reversibility of aggregation of the α -amylase folding intermediate under cytosolic conditions, we studied the stability of aggregates resuspended in a solution of foreign proteins under physiological conditions of pH, ionic strength and temperature. We postulated that protein-protein interactions could destabilize the aggregates. The results obtained (Fig. 7) confirmed this hypothesis and showed that bovine serum albumin, a protein known to bind a variety of hydrophobic compounds [12], was the more efficient effector of disaggregation.

DISCUSSION

We have shown here that the α -amylase folding intermediate is prone to form insoluble aggregates. The ratio of aggregation may be controlled by an initial dimerization step, since the order of the reaction is close to 2.

The α -amylase monomers that make up the aggregates were slightly different in their secondary structures from their soluble counterparts. The aggregation process, however, caused the protein to become resistant to protease. One of the most striking features of the aggregates is their instability under conditions that mimic the cytosolic environment. Protein-protein interactions under mild conditions of pH, ionic-strength and temperature promote disaggregation. One can thus ask the question whether aggregation of partially folded protein is always an irreversible process *in vivo*. It is currently accepted that under such an environment, the chaperonin system prevents aggregation [13] by shielding the hydrophobic surfaces of unfolded proteins. GroEL and GroES, the two components of the chaperone machinery, are present in the *B. subtilis* cytosol [14]. However, the fact that the cellular interior is very crowded with proteins suggests that there are strong protein-protein interactions which might affect the reaction rates and equilibria of many macromolecular reactions [15].

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Table 1. Reaction orders and half-times of the aggregation of the α -amylase refolding intermediate.

Kinetic data were analyzed by non linear least squares fitting to the integrated rate equation given in the text (the Sigma Plot curve fitting programme).

% TFE order (α)	reaction	half-time (s)
0	1.81 ± 0.04	960 ± 35
10	1.82 ± 0.05	274 ± 15
15	1.89 ± 0.09	145 ± 6

Table 2. Secondary structure of α -amylase in native state and aggregated state.

State	Peak position (cm^{-1})	Assignments	% of secondary structure
Native	1633 1687	β -sheets	50
	1652	α -helices	34
	1660-1680	other structures and turns	16
	Refolding intermediate aggregated	1618 1687	intermolecular β -strands
1641		β -sheets	43 (27.1) ^a
1656		α -helices	7 (4)
1671		turns	20 (20.8)

^a Numbers in parenthesis refer to the secondary structure of the refolding non-aggregated form of the intermediate of α -amylase evaluated *via* U.V. CD spectroscopy [1].

Figure 1

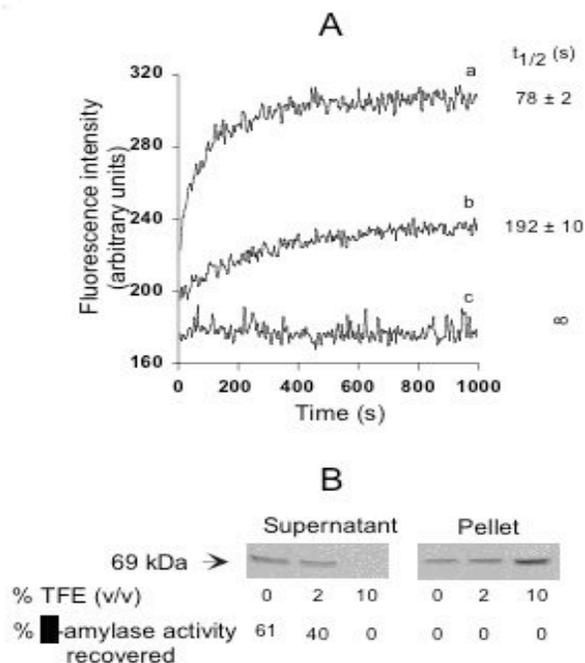


Fig. 1. Competition between folding and aggregation of the α -amylase refolding intermediate in the presence of increasing concentrations of TFE.

The refolding intermediate was obtained by a two-step procedure: α -amylase was first totally unfolded by mixing 1 μ l of α -amylase stock solution (60 mg/ml) with 5 μ l 6 M Gdn-HCl. After 10 min incubation at 37°C, the denaturing mixture was then diluted with 0.5 ml 0.1 M sodium phosphate at pH 7 containing 0.1 mM EDTA. The refolding intermediate formed almost instantaneously in such conditions [1]. Folding of the intermediate was initiated by adding 0.5 ml phosphate buffer containing 1 mM CaCl₂ and 4 % or 20 % TFE at 37°C.

A) Changes in fluorescence intensity were recorded at 344 nm, the excitation wavelength was 280 nm. The dead time for mixing was approximately 8 s. The final concentration of TFE in the refolding reaction mixture were 2 % (b), and 10 % (c). (a) was a control without TFE. The half time of refolding ($t_{1/2}$) for each condition is indicated on the right of the refolding curves.

B) Partitioning between the soluble native state and the aggregated state was measured by enzymatic assay and SDS-PAGE analysis. After recording the fluorescence intensity, the samples were centrifuged (12000 g, 15 min) and the protein pellets resuspended in 1 ml 0.1 M sodium phosphate at pH 7 containing 0.5 mM CaCl₂. Enzyme activity was measured with 25 μ l supernatant and 25 μ l resuspended aggregates from each sample, and SDS-PAGE was performed using 60 μ l aliquots.

Figure 2

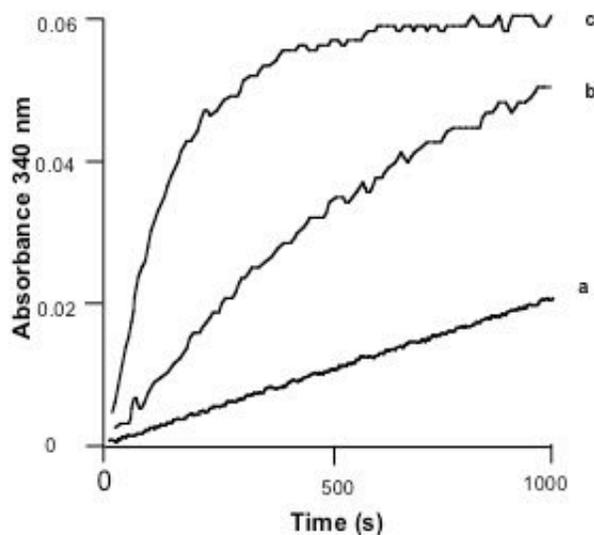


Fig. 2. Kinetics of the aggregation of α -amylase refolding intermediate monitored by the increase in light scattering at 340 nm.

The refolding intermediate was prepared as described in the legend to Fig. 1. Aggregation was initiated by mixing 0.5 ml refolding intermediate solution with 0.5 ml of 0.1M potassium phosphate at pH 7, containing 0.1 mM EDTA and TFE. The final concentration of TFE in the refolding reaction mixture was 15 % (c) or 10 % (b). (a) was the control without TFE. The final concentration of protein in the aggregation mixtures was 60 μ g/ml. Changes in light scattering at 340 nm were monitored using a Unicam spectrometer thermoregulated at 37°C. The dead time was 8 s.

Figure 3

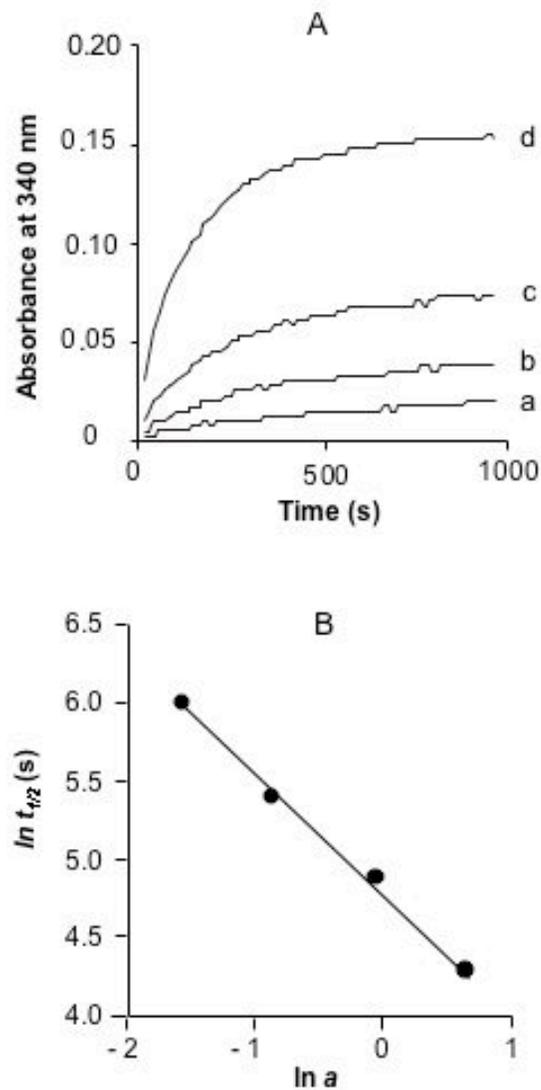


Fig. 3. Kinetics of aggregation of the refolding intermediates of α -amylase in 15 % TFE at various initial protein concentrations.

A) The refolding intermediate was prepared as described in the legend to Fig. 1. The amounts of protein subjected to unfolding were calculated so that the final concentrations of the refolding intermediate in the aggregation mixture were 15 $\mu\text{g/ml}$ (a), 30 $\mu\text{g/ml}$ (b), 60 $\mu\text{g/ml}$ (c) and 120 $\mu\text{g/ml}$ (d).

B) Plot of $\ln a$ against $\ln t_{1/2}$.

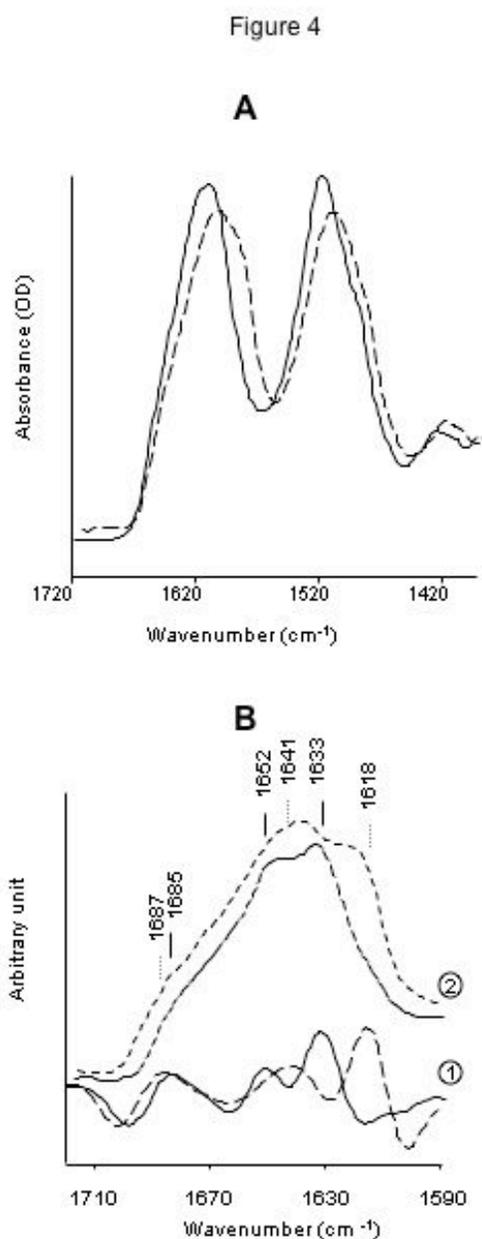


Fig. 4. ATR IR spectra of α -amylase in the native state (black line) and folding intermediate aggregates (dotted line).

The refolding intermediate was prepared as described in the legend to Fig. 1. Aggregation was carried out as described in the legend to Fig. 3. The final concentration of the refolding intermediate in the aggregation mixture was 60 $\mu\text{g/ml}$. Aggregates were centrifuged for 20 min at 12 000 g and washed with buffer only. The pellet corresponding to 60 μg protein was resuspended in 2 μl 0.1 M potassium phosphate at pH 7. For the control, 1 μl α -amylase stock solution (60 mg/ml) was mixed with 1 μl 0.1 M potassium phosphate at pH 7. Samples were deposited on the ATR crystal.

A) Normalized absorbance spectra

B) Second derivative amide I (1) and self deconvolution (2) spectra derived from A.

Figure 5

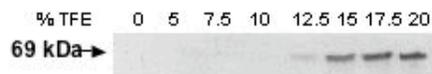


Fig. 5. Kinetic competition between aggregation and proteolysis of the refolding intermediate of α -amylase.

α -amylase (2 μ l stock solution) was mixed with 12 μ l 6 M Gdn-HCl. After 10 min at 37°C, the refolding intermediate was stabilized by diluting the denaturing mixture in 1 ml 0.1 M sodium phosphate at pH 7 containing 1 mM EDTA at 37°C. A set of Eppendorf tubes containing subtilisin and TFE was prepared. Drops of each solution were added separately in the bottoms of the tubes. The final concentration of subtilisin was increased as the TFE concentration increased to counterbalance inhibition of subtilisin by TFE. Aliquots of refolding intermediate solution (100 μ l) were rapidly introduced into each tube, and quickly mixed with the TFE and subtilisin. After 5 min incubation, 0.1 M PMSF (5 μ l) was added and the samples were analysed by SDS PAGE.

Figure 6

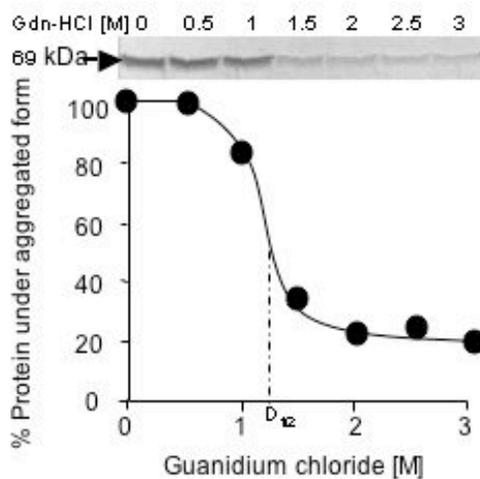


Fig. 6. Solubilization of α -amylase aggregates by Gdn-HCl.

The totally unfolded protein, obtained from incubating 2 μ l stock solution with 8 μ l 6 M Gdn-HCl, was diluted in 1.7 ml 0.1 M sodium phosphate at pH 7 containing 0.5 mM EDTA. Aggregates were formed by adding 0.3 ml TFE. The suspensions of aggregates were divided into 10 fractions of 0.2 ml each and centrifuged (20 min, 12000 g). The supernatants were discarded and the pellets (approximately 12 μ g protein) were resuspended in 0.1 ml 0.1 M sodium phosphate buffer at pH7 containing 0.5 mM EDTA and increasing concentrations of Gdn-HCl and then incubated at 37°C for two hours. The samples were centrifuged and the supernatants and pellets were analysed by SDS PAGE. A-amylase in the soluble phase and pellet was quantified by densitometry of Coomassie blue stained gels using the NIH Image program. Prolonged incubation in Gdn-HCl did not significantly modify the result.

Figure 7

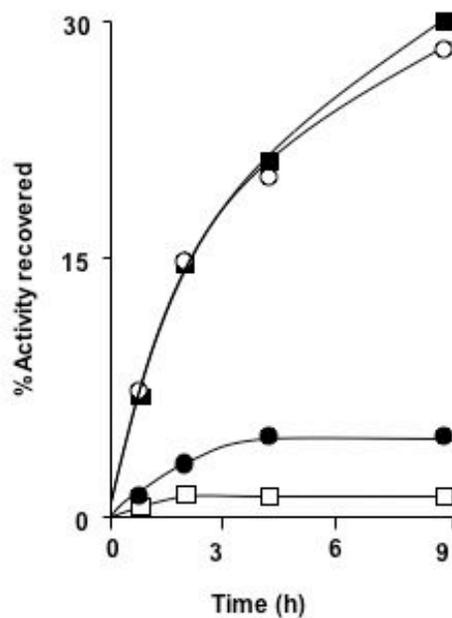


Fig. 7. Effect of foreign proteins on the stability of α -amylase aggregates.

Aggregates of α -amylase refolding intermediates were prepared as described in the legend to Fig.4. The aggregates were resuspended in 0.1 M sodium phosphate buffer at pH7 containing 0.5 mM Ca^{2+} and 1 mg/ml of BSA (■), ovalbumin (○) or lysozyme (●). Control without protein (□).

The suspensions were incubated at 37°C and 50 μl samples were withdrawn and centrifuged at intervals. The supernatants were mixed with 0.25 ml 0.1 M sodium phosphate buffer at pH 7 containing 0.5 mM Ca^{2+} for 15 min, and the enzyme activity was assayed. The recovery of active soluble enzyme activity was evaluated with respect to the amount of unfolded enzyme used to prepare the aggregates.