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Study on Vegetable Field Evaluation Index System for Non-point Source Pollution of Dagu River Basin

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Abstract. This research combined data from soil, terrain and meteorological surveys, with quantitative data from soil and water field studies to determine the leaching processes of nitrogen and phosphorus contaminants from field vegetation into the Dagu River. Using the Gray correlation analysis method, precipitation and irrigation were established as the major driving force behind leaching losses from cultivated land. The type of fertilizers used, and the chemical and physical properties (porosity and bulk density) of soils in the region were set as parent sequences. COD values of nitrogen and phosphorus concentration were calculated from soil and water samples. Also, the depth of core samples was factored into the analysis. The result showed that the supply of soil nitrogen and phosphorus, which were mainly derived from fertilizers, were not generally high levels in Qingdao. According to fuzzy association degree, this investigation confirmed main influencing factors which affected COD, total nitrogen, total phosphorus of soil and groundwater, ammonia nitrogen and nitrate nitrogen of groundwater respectively. And according to these factors the initial vegetable field evaluation index system for non-point source pollution of Dagu River basin was constructed.

Key words: Gray correlation analysis, Nitrogen, Phosphorus, COD, Precipitation, Soil physical properties, Soil type, Index system

1. Introduction

Excessive nitrogen and phosphorus values used in fertilization will significantly decrease the economic benefits of enhanced crop production, as well as increase the risk of nitrogen and phosphorus contaminants in water reserves [1-3]. Approximately, 60% of water pollution is caused by non-point source (NPS) pollution in America [4]. In Northern Australia, NPS pollution flowing into water reserves is also the major source of nitrogen contamination [5]. In Denmark, 94% of the nitrogen load and 52% of the phosphorus load in 270 rivers is a result of non-point source processes [6]. The same effects can be seen in the Netherlands, where 60% of the total nitrogen contaminants and 40-50% of the total phosphorus contaminants are caused by NPS pollution [7].

In recent years, as agricultural development mirrors population growth in China, NPS pollution has become the main source of water pollution. Currently, China contains 1600 million hm^2 of cultivated land, which amounts to 35% of the agricultural land on the planet [8]. Statistics measuring the amount of fertilizers, livestock and poultry manure and the cultivated land area from 2000 to 2006 in Qingdao were investigated. Subsequently, an index was established to evaluate the nitrogen and phosphorus content in livestock manure as

NPS pollutants [9]. A Grey Correlation study to evaluate water pollution from vegetable fields in the Dagu River Basin of Qingdao[10].

Over the years, the Dagu River has become increasingly more important to the national economy and to the lives of local residents. However, as the region surrounding the Dagu River Basin is developed, pollution of the river has become increasing obvious. It is the fertilizers and pesticides that have aided the agricultural development that threaten the water reserves. The numerous and excessive applications of fertilizers and pesticides cause soil erosion and serious water pollution from increasing nitrogen and phosphorus concentrations. This study integrates data from soil, terrain and meteorological surveys of Dagu River basin, with quantitative analysis studies of thousands of soil samples to construct an indexing system of NPS pollutants entering the Dagu River. The Dagu River is located in the western Shandong Peninsula, between $120^{\circ} 03' \sim 120^{\circ} 25'E$ and $36^{\circ} 10' \sim 37^{\circ} 12'N$ and covers an area of 4631.3 km^2 in Qingdao. The river runs through five districts in Qingdao, Laixi City, Pingdu, Jiaozhou, Chengyang, Jimo (see Figure 1). The annual precipitation in the region is about 685.3 mm and an annual runoff is about $6.311 \times 10^3 \text{ m}^3$ [11].

1. Materials and Methods

1.1. Sample Collection

For this study, soil and water samples were collect in December of 2009 from cultivated land throughout the region (Figure 1). The samples of vegetable field soil were divided into two categories, surface soil (0-20 cm) and submerged soils (80-100 cm). Additionally, surface water and ground water samples were collect from the same areas. GPS was used to map the latitude and longitude of sample areas. All samples were analyzed in the laboratory, using quantitative methods.

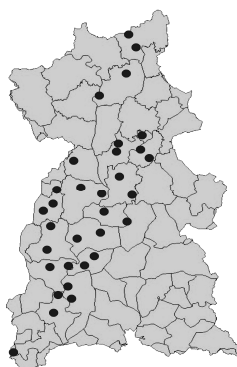


Fig.1. Soil samples distribute

1.2. Sample Analysis

To determine the COD of water samples, they were heated in a strongly acid solution, containing potassium dichromate. A Euro Tech ET3150B multiple digester and an ET1151M COD Monitor were used for characterization. $\text{K}_2\text{S}_2\text{O}_8$ digestion molybdenum blue colorimetry was used alongside the COD digester to analyze the total concentration of phosphorus in samples. Nitrite has a visible absorption at 220nm , so a 752SP UV was used to measure its concentration in aqueous samples. Nessler's reagent was added to

pretreated samples, and, then, a spectrometer was used to measure their absorbance at 420 nm. Also, oxidation-ultraviolet spectrophotometry was used to determine the total nitrogen content of samples, after the addition of potassium persulphate at 120~124°C.

Soil samples were first treated with alkaline potassium persulfate. Then, UV spectrophotometry was used to measure each solutions absorbance at 220nm, 275nm and 700nm to measure their total nitrogen and phosphorus content. Potassium dichromate was added to soil samples before COD analysis.

2. Results and Analysis

2.1. The main source and loss of non-point source pollution

Fertilizer, irrigation, and atmospheric deposition are the main sources of nitrogen and phosphorus in cultivated land within the basin. Among these, fertilizers are the most abundant source of the elements. There was a large quantity of NO_3^- -N in samples, as it is not readily absorbed by soil colloid [12]. Rainfall and irrigation contributed to the loss of accumulated NO_3^- -N, via surface run-off (procedure C in figure 2) and/or through leaching (procedure A and B in figure 2) [13],[14].

2.2. The substance basis of fertilizer leaching and driving force of vegetable field soil nitrogen and phosphorus

A classification system was developed from the analysis of more than 7000 soil samples, which were classified on the bases of total nutrient content. Results were classified into six categories based on nitrogen supply level from vegetables. The data is shown in table 1.

Table 1. Classification total nitrogen in soil (Data from Qingdao soil, edited by the Soil and Fertilizer Workstation of Qingdao City)

supply level	field			
	Upland area		Vegetable land	
	acre	%	acre	%
high	—	—	—	—
extremely high	5938	0.07	10174	3.16
general	1060672	12.14	211928	65.78
lower	4830122	55.31	100042	31.06
low	2676373	30.65	—	—
extremely low	160226	1.83	—	—

We analyzed more than 1000 soil samples and classified the soluble phosphorus content. Results showed that very high soluble phosphorus supply level of vegetable land was only 7.83%, but general and lower supply level were 76.99% (Table 2).

Table 2. Classification soluble phosphorus in soil (Dadas from <Qingdao soil >edited by the Soil and Fertilizer Workstation of Qingdao City)

Supply level	field			
	Upland area		Vegetable land	
	acre	%	acre	%
high	—	—	—	—
higher	—	—	25207	7.83
general	277110	3.17	147990	45.93
lower	1797677	20.58	129624	40.24
low	3419031	39.15	19323	6
extremely low	3239513	37.1	—	—

From the above statistics, we found that nitrogen and phosphorus supply level in Qingdao vegetable land were not very high. Additionally, fertilizer was the main source of nitrogen and phosphorus. Subsequently, it was reasonable to study fertilizer as a basic source of fertilizer leaching. Meteorological precipitation data combined with crop irrigation were overwhelmingly contributed to the leaching loss of nitrogen and phosphorus from soil.

2.3. Determine leaching loss factors of nitrogen and phosphorus according to grey relational analysis

Types of fertilizer and soils, the physical properties of different soils (porosity and bulk density), precipitation data, soil nitrogen and phosphorus content, and soil COD were indexes in the analysis of leaching nitrogen and phosphorous. Gray correlation was used to analyze the effect of these indexes on the contaminant concentrations in the local water reserve. An index system based on well-correlated factors related to nitrogen and phosphorous loss from agricultural soil was established. The factors are listed below, in decreasing order: ammonia content, nitrate content, total nitrogen of groundwater, soil type, soil porosity (from shallow to deep), soil bulk density (from shallow to deep), precipitation, soil COD and soil total nitrogen content. Fertilizer plays the most significant role in affecting the quality of groundwater, followed by soil physical properties and soil chemical properties. The order of factors affecting total phosphorus content of groundwater was soil type, soil phosphorus content, soil COD, phosphate fertilizer of per hectare, soil porosity (from shallow to deep), soil bulk density (from shallow to deep) and precipitation, respectively. Phosphorus leaching from soil was the main source of phosphorus in groundwater.

2.4. Single factor assessment index of non-point source pollution

2.4.1. Fertilizer grade of nitrogen and phosphorus leaching loss.

Compound fertilizer, with similar nitrogen and phosphorous contents commonly used fertilizers, was applied to survey region (total nitrogen content 15% according to N, total phosphorus content 15% according to P₂O₅). The result from formula (1, 2) gave a mean value of fertilizer in survey region of 150.7 kg/hm². The standard deviation was 139.6.

Mean value:

$$\bar{x} = \frac{\sum_{i=1}^n x_i}{n} \quad (1)$$

Standard Deviation:

$$\sigma = \sqrt{\frac{\sum_{i=1}^n (x_i - \bar{x})^2}{n}} \quad (2)$$

Probability Density:

$$f(x) = \frac{1}{\sqrt{2\pi} \cdot \sigma} e^{-\frac{(x-\bar{x})^2}{2\sigma^2}} \quad (3)$$

According to probability density calculations (figure 2), the second center should be considered as the average nitrogen concentration in fertilizer per hectare in every season. The standard deviation was divided into five intervals: 0~80, 80~220, 220~360, 360~500 and >500 kg • hm⁻² • season⁻¹ (see table 4). An advantage of this division was that similar values can be drawn in same level, which reduces the risk that similar dispersed within the data.

Table3. Sequence of gray correlationTable3a

	Groundwater		
	Ammonia Nitrogen	Nitrate Nitrogen	Total Nitrogen
Nitrogen fertilizer	1	1	1
Soil type	2	2	2
Soil porosity			
shallow layer	3	3	3
medium layer	4	4	4
deep layer	5	5	5
Soilbulk density			
shallow layer	6	6	6
medium layer	7	7	7
deep layer	8	8	8
Precipitation	9	9	9
COD of soil	10	10	10
Soil total nitrogen	11	11	11

Table3b

TP of groundwater	
soil tvpe	1
Soil total phosphorus	2
COD of soil	3
Nitrogen fertilizer	4

Porosity (deep layer)	5
Porosity (medium layer)	6
Porosity (shallow layer)	7
bulk density (shallow layer)	8
bulk density (deep layer)	9
bulk density (medium layer)	10
Precipitation	11

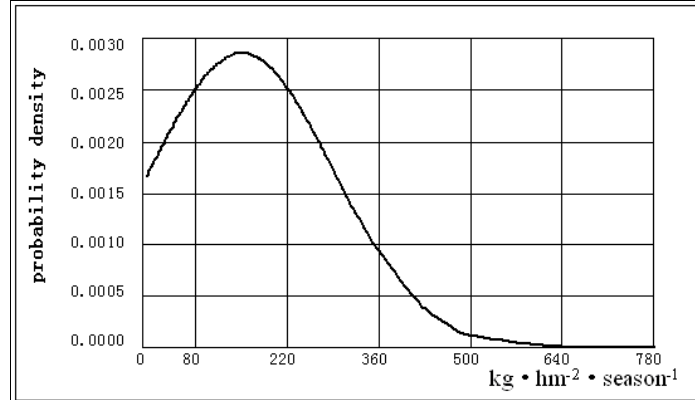


Fig.2. Probability density function of fertilization

Table 4 Fertilizer grade of nitrogen and phosphorus leaching

Compound fertilizer ($\text{kg} \cdot \text{hm}^{-2} \cdot \text{season}^{-1}$)	Grade	Description
<80	F1	Fertilizer less than the average fertilizer level. The substance basis of leaching loss was not enough, and the leaching loss won't be happened.
80~220	F2	Fertilizer was near the average amount. On the occasion of heavy rain or over-irrigation, light leaching loss will be happened.
220~360	F3	The light leaching loss will be happened. The degree will increase according to rate of fertilizer and water.
360~500	F4	Middle leaching loss level. But heavy leaching loss can be happened. The rate of fertilizer is more than average rate.
>500	F5	(Super) heavy leaching loss level.

2.4.2. The precipitation grade of nitrogen and phosphorus leaching loss.

When long-term soil leaching occurs, the optimum conditions were as follows: the effects of the sum of precipitation and irrigation are more than the effects of the sum of run-off, evaporation and good soil infiltration. Conditions of short-term soil leaching mirrored large amount of precipitation or irrigation [15]. In the study area, precipitation is heaviest from June to August, about 500mm, which accounts for more than 50% of the annual precipitation. In the summer irrigation was lessened, due to the increase precipitation.

In this analysis, 200mm precipitation was used as the lower limit and was divided into 5 levels in 100 mm intervals. Short-term soil loss reached measurable level when single precipitation/irrigation events exceeded 50mm (see the table 5).

Table 5 Driving factor grade of nitrogen and phosphorus fertilizer leaching

Precipitation (mm. season ⁻¹)	grade	description
<200	W1	Shortage of precipitation. The general leaching loss won't be happened.
200~300	W2	Precipitation met the requirements of leaching loss. Light leaching loss may be happened.
300~400	W3	Precipitation met the requirements of leaching loss. Light leaching loss may be happened.
400~500	W4	Precipitation was adequate, and moderate leaching loss may be happened.
>500	W5	Precipitation was extremely adequate, and heavy leaching loss may be happened.

2.4.3 The soil type grade of nitrogen and phosphorus leaching loss.

Soil type played an important role in leaching loss. Under similar conditions, clay soil showed minimum nitrogen and phosphorus leaching loss, whereas, the largest nitrogen and phosphorus leaching loss was from sand soil. Medium loss was observed in loam soil [16],[17]. The soils of Dagu River, which can be used to cultivate vegetables, are brown earth, aquic brown earth, brown paddy soil, cinnamon soil, eluvial cinnamon soil, developed cinnamon soil, lime concretion black soil, and fluvo-Aquic soil. Nitrogen and phosphorus leaching loss was divided into three categories, according to an analysis of the physical property of the soils, including total soil porosity and soil bulk density (see table 6).

Table 6 The soil type grade of nitrogen and phosphorus leaching loss

Soil type	grade	description
brown earth	—	It widely distributes in the hill, valley and the front slope of mountain. The land is thick, and there is clayey layer generally. It played reduced role on nitrogen and phosphorus leaching loss.
aquic brown earth	—	High degree of maturation, viscous soil. It played reduced role on nitrogen and phosphorus leaching loss
brown paddy soil	+	Thin layer, rough texture, high impurity content. Lower capability of moisture and fertilizer conservation. An enhanced nitrogen and phosphorus leaching loss type.
cinnamon soil	—	The process of sticky soil obviously, soil deep. Higher capability of moisture and fertilizer conservation. It played reduced role on nitrogen and phosphorus leaching loss.
Eluvial cinnamon soil	—	Distribution in the slope and valley. Thick layer. Higher capability of moisture and fertilizer conservation. It played reduced role on nitrogen and phosphorus leaching loss.
developed cinnamon soil	+	Thin layer. Gravel. An enhanced nitrogen and phosphorus leaching loss type.
lime concretion black soil,	—	Thick layer. Sticky. It played reduced role on nitrogen and phosphorus leaching loss.
fluvo-Aquic soil	/	Different physical and chemical properties and different degree of reposado. The soil type grade of nitrogen and phosphorus leaching loss was unclear.

2.4.4 The soil body configuration grade of nitrogen and phosphorus leaching loss.

According to results of table 5, when sandy soil is present in the upper layer, leaching will occur more readily. On the contrary, when clay soil is present, leaching will be respectively lower. Consequently, we classified the nitrogen and phosphorus leaching loss according to the soil composition. Areas of sand-layered and thin-layered soil were classified as enhanced nitrogen and phosphorus leaching loss types. Clay layered, intercalated clay layered, Mengyu type, and Mengyin type soils were as classified as reduced nitrogen and phosphorus leaching

loss types.

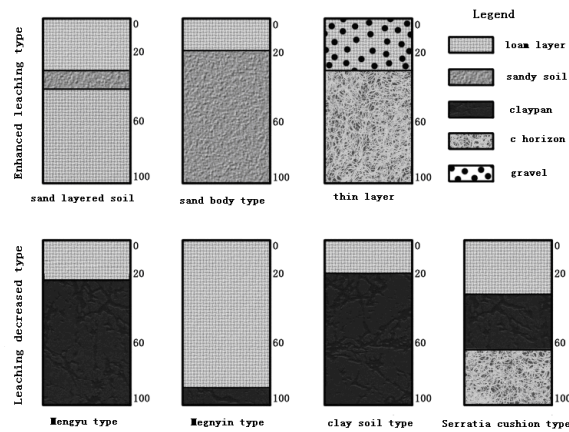


Fig. 3. The Soil Body Configuration of nitrogen and phosphorus leaching loss

3. Conclusions

Fertilizer was as the main source of nitrogen and phosphorus NPS pollution. Precipitation was the most significant driving force of leaching loss of nitrogen and phosphorus in regional soils. The application of fertilizer is the most significant contributor to nitrogen content in groundwater, followed by the physical and chemical properties of regional soils. The results showed that the most influential factors determining the total nitrogen content of groundwater was the soil type, chemical properties of the soil, phosphorus content, the physical properties of the soil, and precipitation, respectively. The amount of nitrogen per hectare and the precipitation in a season were divided into five levels respectively. Nitrogen and phosphorus leaching loss per soil type was classified as follows, enhanced grade, reduced grade and uncertain grade. Soil body configurations of nitrogen and phosphorus leaching loss were classified as enhanced and reduced grades.

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